Which solid contains more than one kind of bonding

A crystalline solid is a solid that consists of particles like atoms, ions, or molecules arranged consistently in a long range order. This pattern repeats throughout the solid, giving it a uniform structure. Most solid substances are ...

The Lewis structures depict an electron where one dot represents one valence electron and the lines connecting the atoms represent one shared pair of electrons, which are also referred to as one covalent bond. This ...

Hydrogen bonding, interaction involving a hydrogen atom located between a pair of other atoms having a high affinity for electrons; such a bond is weaker than an ionic bond or covalent bond but stronger than van der Waals ...

The Lewis structure of H2SO4 shows the molecular arrangement of sulfuric acid, a compound with two hydrogen, one sulfur, and four oxygen atoms. In this structure, there are four single bonds and two double bonds. The ...

Web: https://ichipcorp.co.za